

NO DRAWINGS

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Date of Application and filing Complete

Specification: 24 May, 1966.

No. 22999/66

1,121,563

Complete Specification Published: 31 July, 1968.

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Index at Acceptance:—C4 X11; A5 B31.

Int. Cl.:—B 05 b 17/00.

COMPLETE SPECIFICATION.

Self-propelled Homogeneous Liquid Compositions

We, YARDLEY AND COMPANY LIMITED, a British Company of 33 Old Bond Street, London, do hereby declare the invention, for which we pray that a patent may be 5 granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:

The present invention relates to the production of self-propelled homogeneous liquid 10 compositions which when dispensed from suitable aerosol containers produce foams of

transient stability.

There have been proposed and used, aerosol foam compositions for cosmetic pur-15 poses. While such compositions have been stated to be suitable for use in transparent containers or dispensers, they have had the disadvantage in that at low ambient temperatures some of the constituents separated 20 out. They formed at the bottom of the aerosol an unsightly heterogeneous mass or precipitate. Such temperatures below 60°-65° F., and often as low as 45° F. are often encountered in the bathroom in mornings

25 of low outside temperatures. These precipitates, usually of the surfactants present, were frequently so serious that foaming has even been prevented at such low peratures.

The present invention is intended and adapted to overcome the defects in the prior art, it being among the objects of the invention to provide an aerosol foam composition which remains clear at said low tem-35 peratures and prevents any noticeable

amount of precipitation in the container. Compositions according to the invention can be made up which will not cause stinging of the skin, avoid producing red blotches

40 on the skin and which will not cause irritation thereof.

The constituents and the proportions thereof can be so chosen that the foam produced therefrom will be self-collapsing in a very short time.

The invention is directed to the production of such fluid compositions at low pressures and with clarity at extremely low ambient temperatures so that the product will remain clear and fluid over a wide tem- 50 perature range and thus be esthetically suitable for packaging in clear glass, plastic coated glass and clear plastic aerosol containers. Body fresheners, after shave lotions, hair grooms, and other similar quick break- 55 ing foams are made considerably more elegant and desirable when dispensed from a clear single phase aerosol package.

Part of the problem has been restricting surfactant selection primarily to certain com- 60 merically available blended surface active agents. In order to achieve low temperature clarity, applicants have had to develop their own special blends of suitable surfactants in addition to making other modifications 65 in composition which will be more fully

described later.

SKIN STINGING AND IRRITATIONAL POTENTIAL

Applicants' work in this area has shown 70 that high alcohol levels (i.e., 55-65%) are quite irritating to the intact skin often causing red blotches upon second application. Since many cosmetic applications of these foams are on the skin, this is quite un- 75 desirable. This effect is even more pronounced in after-shave foams which are applied to abraded skin.

For these reasons Applicants have developed foams with ethanol contents as low 80 as 32% of the 95% alcohol, or even lower in some cases. Even non-irritating alcohol levels may sting the face and therefore cosmetic elegance demands foams of negligible stinging potential. Collapsible foam products 85 of the present invention formulated with

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alcohol levels in the range of 30-45% alcohol possess no or minimal stinging attributes. FOAM DURATION AND COLLAPSIBILITY

Another aspect of the present systems is the stability of the foam puff. Prior art compositions were often times too stable. As distinctly different from the prior art, Applicants have found that suitable foams should 10 be of the self-collapsing type with durations of from 10 to 60 seconds. Consumers have actually shown a preference for this type since more stable foams resist rubbing and are often-times extruded between the fingers 15 when handled. The foam puff characteristics are controlled by the concentration and composition of the surfactant blend, the concentration of the alcohol, and the type and concentration of propellants selected.

20 INGREDIENTS

The essential basic ingredients of a quickbreaking foam of transient stability are as follows:

A-A blend of alcohol (30 to 55%) and

25 water to act as the vehicle.

B-A blend of surfactants at concentra-

tions from 0.4 to 5.0% by weight.

C—Either one or a blend of compatible hydrocarbon or halocarbon gases at con-30 centrations from 2 to 18% by weight.

Additionally, any useful auxiliary diluent might be employed which would not interfere with the clarity or the foam of the finished aerosol composition.

35 A-The Alcohol/Water Ratio

The surfactants normally used in these aerosols are chosen to be soluble in the alcohol and insoluble in water. The water/alcohol concentration ratio is then adjusted ambient temperature to the point at which

40 at ambient temperature to the point at which the surfactant blend is just barely insoluble in the mixture. The temperature at which the concentrate will clear and become homogenous usually coincides with the upper

45 temperature at which a usable foam may be generated when the concentrate is dispensed as an aerosol.

The expression "alcohol" refers to ethanol, methanol, isopropanol or n-propanol.

50 When lower alcohol levels are used, stinging of the skin is reduced, therefore Applicants have endeavored to employ as little alchol as possible in the composition concomitant with clarity of the finished 55 aerosol. This is accomplished by:

1. Selection of surfactants with good

hydro-alcoholic solubility.

2. Use of glycols and ethers as couplers.

3. Increased propellant concentration.

60 B-Surfactant Selection

The character of the foam and its stability are most directly affected by the concentration and choice of surfactant. In no case however, should it be necessary to use more 65 than 5% by weight of surfactant in the total

aerosol because additions in excess of this quantity produce tackiness, and insulated feeling on the skin and are only really acting as solubilizing replacements for alcohol. This replacement for alcohol may be more elegantly achieved by using more propellant, glycol, or water, or a combination of these, thus avoiding deposition of excessive solids on the skin. Suitable surfactant blends are somewhat limited because they must be 75 solid substances, clearly soluble in the finished aerosol composition including propellants down to about 6° C., but insoluble in the concentrate (all the remaining ingredients except propellants) at temperatures 80 as high as 30° C., or even higher, if foam is to be generated at temperatures higher than 30° C.

25-60 Cetyl Alcohol pentaethoxy (5 moles ethylene 95 oxide) C₁₆₋₂₀ Fatty Alcohol ether 75-40 33-66.0 Cetyl Alcohol diethoxy Stearyl ether 53-20 20-14 deca ethoxy Stearyl ether *5*0.0 100 Cetyl Alcohol polyethylene glycol (40 moles ethylene oxide lanolin alcohol 50.0 ether 28-35 diethoxy Stearyl ether . 105 Cetyl Alcohol 3- 6 Polyethylene Glycol (25 moles ethylene oxide) lanolin alcohol 58-66 ether 6.0 Arachidyl Alcohol Polyoxyethylene (40 moles 110 ethylene oxide) lanolin alcohol 71.0 ether Diethoxy Stearyl ether 23.0 54-64 diethoxy Stearyl ether 115 46-36 Cetvl Alcohol

Blends of nonionic surfactants based upon ethers of sorbitan fatty acid esters, ethoxylated, saturated C₁₅₋₂₀ fatty alcohols and ethoxylated lanolin alcohols have also been found to be extremely effective surfactants 120 when good foam and low temperature clarity is desired in systems containing lower alcohol levels. Examples of such systems are as follows:

g) Any or a blend of Penta ethoxy-decaethoxy Cetyl/Stearyl/Arachidyl ethers 100.0
h) Diethoxy Stearyl Ether 25-48
Polyoxyethylene (40 moles 130

1,121,563

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		ethylene oxide) Lanolin	an
		alcohol ether 75-52	str
		i) Tetra ethoxy Stearyl ether 25	
		Polyoxyethylene (20 moles	ing
	5	ethylene oxide) Sorbitan	pe
	-	monolaurate 75	WC
		j) tetraethoxy Cetyl/Stearyl ether 35-40	
		decaethoxy Cetyl/Stearyl ether 65-60	otl
		C—The Propellants	be
	10	The solubility of propellants in high	sti
		water systems is a limitation on the pro-	tio
		pellant concentration. To overcome this	ag
		limitation we select propellants which have	an
		the highest relative water solubility. Ex-	suc
	15	amples of some of these propellants are	mi
		butane, monochloro-difluoro-ethane, di-	
		fluoro-ethane, trichloro-monofluoro-methane	tra
		and dichloro-monofluoro-methane which	the
		help to increase propellant compatibility	
	20	with the concentrate.	
		These more soluble propellants may often-	95
		times be used with other less soluble gases	Su
		such as dichloro-difluoro-methane and tetra-	Gl
		fluoro-dichloro-ethane to derive the best	W
	25	characteristics of both types.	Ce
		All other things being equal, the maxi-	Pe
		mum concentration of propellant which the	W
		system can maintain in a single liquid phase	pre
	••	is dependent upon the level of water in the	COI
Ļ	30	aerosol. Less water in the concentrate allows	ger
		more propellant to be dissolved but the	fo
		foam characteristics are also dependent	bu
		upon the water, and poor foam generation	wh
	25	results if the water concentration is reduced	pe
	دد	below 20% unless high glycol levels (above	aeı
		10%) are utilized to make up for the lower	as
		Water percentage.	
		The lowest pressures are obtained in low water systems containing monochlor-di-	
	40	water systems containing monochlor-di- fluoro-ethane or a combination of trichloro-	
	₩	monefluoro-methane or tetrafluoro-dichloro-	
		ethane in about equal proportions with said	
		monochloro-difluoro-ethane. Low pressures	C
		may also be achieved with blends of di-	_
	45	chloro-difluoro-ethane with trichloro-mono-	
		fluoro-methane or tetrafluoro-dichloro- ethane containing 30% or less by weight of	
		dichloro-difluoro-methane. The pressures	
		which are preferred range from about 15 to	
	50	25 psig, but under some conditions pres-	
		sures may range from 10 to 40 p.s.i.g.	P
		Propellant solubility is also critical be-	
		cause foam stability is a function of the	Th
		solubility of the propellant in the aqueous	gre
	55	alcohol solution.	aι
		D—Auxiliary Ingredients (Couplers,	aeı
		diluents, germicides, etc.)	the
		While the basic essential ingredients of	bo
		these improved foams have been outlined as	mi
	60	consisting of alcohol, water, propellants and	the
		a nonionic surfactant blend, other useful	sin
		ingredients such as glycols, fatty ethers, and	6°

ingredients such as glycols, fatty ethers, and fatty esters may be included as additional couplers or emollients. Certain liquid surf-

65 actants such as di- and tri-ethoxy oleyl ether

and tetra-ethoxy lauryl ether, though not structural foaming agents, assist in produc1,121,563
ing clarity down to lower temperatures or permit use of higher propellant levels than 70 would otherwise be attainable.

Healing agents, bacteriostatic agents and other medicaments and emollients may also be included for their utility but do not constitute a critical part of the present invention. In any case low stinging auxiliary agents such as these may also be selected and anti-irritants and anti-inflammatories such as polyvinyl pyrollidone or guaiazulene might also be included.

The following specific examples are illustrative of the invention although not limiting the scope thereof:

Example 1

9	by Wt.	85
95% Ethanol	32.60	
Surfactant blend "g"	0.40	
Glycerine	10.00	
Water	56.81	
Cetyl trimethyl ammonium bromide		90
Perfume	0.14	70
When two grams of diffuoro-etha	ne was	
pressure filled into the bottle contain	ring this	
concentrate it cleared and became	homos	
geneous. The finished aerosol, fitted	with a	05
foam actuator, produced a goo	d tight	,,
bubbled quick-breaking after-shav	a teem	
which could be stiming and star	e ioam	
which caused no stinging even w	nen re-	
peatedly applied to the face. The	finished	
aerosol was clear at temperatures of	f as low	100
as 6° C.		
- · ·		

	Example 2		
	(95% Alcohol	% by Wt. 51.0	
	(Surfactant Blend "b"	2.7	105
C	(Menthol	0.1	
	(Perfume	0.8	110
	(Hexachlorophene	0.1	
P	(Distilled Water (Trichloro-monofluoro-methane	34.9 4.4	115

(Monochloro-difluoro-ethane 6.1) The above after-shave foam concentrate ingredients (C) were blended together to form a uniform gelled slurry and filled into clear 120 aerosol bottles. After purging to remove air the bottles were scaled with a regular glass bottle aerosol valve. When the propellant mixture (P) was added through the valve, the finished aerosol became a homogeneous 125 single liquid phase clear down to about 6° C. The pressure of the finished aerosol was 20 psig ± 2 psig at 70° F. When applied to the face after shaving, the foam produced no irritation and only the desired 130

minimal momentary stinging sensation.

Example 3

A medicated body freshener foam was prepared according to the formula shown below, 5 introduced into a clear aerosol bottle and sealed with a standard glass bottle valve after first purging to remove air. The heterogeneous concentrate was clarified by the addition of the propellant phase (P). The 10 finished aerosol had a pressure of 25 psig ± 2 psig at 70° F. and was clear down to about 6° C.

When fitted with a standard foam actuator, a foam puff dispensed from the aerosol 15 had desirable foam body and texture and a stability of from 30 to 60 seconds at ambient temperature.

When applied to the skin, no irritation or stinging was produced even after repeated

20 re	-аррисацоп.	% by Wt
	(95% Ethanol	38.70
	(Surfactant Blend "a"	1.90
25	(Hexachlorophene	0.30
С	(Polyethylene Glycol 200	9.70
30	((Perfume	0.2
	(Distilled Water	45.9
P	Monocloro-difluoro-ethane Example 4	.3.3

35 The following medicated skin product concentrate (C) was prepared as in previous Examples and was a semi-solid opaque mass at 70° F. Addition of the propellant blend as before resulted in a clear homogenous 40 aerosol with a pressure of only 18 psig ± 2 psig at 70° F. The entire aerosol packed in a clear uncoated bottle was clear at temperatures down to 6° C. ± 1° C. The foam puff was particularly fine bubbled and emol-45 lient due to the presence of a somewhat higher concentration of surfactant blend.

_ •	-	% <i>by W</i>
	(95% Ethanol	47.8
50	(Surfactant Blend "d"	2.9
	(Water	38.6
С	\	
55	N-lauryl-colamino formyl methy	Al
-	(pyridinium chloride	0.1
	(Perfume	0.9
60	((Allantoin	0.1
	(Trichloro-monofluoro-methane	4.4
P	(
•	(Monochloro-difluoro-ethane	6.1
	Among the advantages inherent	t in th

Among the advantages inherent in the 65 present invention are the absence of stinging

of the skin upon repeated application of the foam aerosol, the absence of irritation of even freshly shaved skin, and the avoidance of formation of red blotches on the skin. An important advantage lies in the clarity 70 of the aerosol at temperatures as low as 6° C. This is obtained by specially selected blends of surfactants, lowering of the alcohol content, increase of the propellant content, and selection of propellant blend.

WHAT WE CLAIM IS: 1. Low pressure aerosol foam compositions consisting of a vehicle, a surfactant blend and a propellant; said vehicle being a monoalcohol having 1 to 3 carbon atoms 80 in the amount of 30 to 55% and water in sufficient amount in the concentrate to make 100%; said surfactant blend being present in amount of 0.4 to 5.0%, soluble at temperatures of about 6° C., in said composition, and being comprised by at least one of the fatty alcohols having 12 to 22 carbon atoms combined with a nonionic surfactant which is a polyethylene oxide derivative of said fatty alcohols, of a fatty acid, of a 90 lanolin acid or of a lanolin alcohol; said propellant being present in amounts of 2 to 18%, and being butane, monochlor-didifluoro-ethane, trichlorofluoro-ethane, monofluoro-methane, dichloro-monofluoro-95 methane, dichloro-difluoro-methane or tetrafluoro-dichloro-ethane, the pressure thereof being 10 to 40 p.s.i.g. at ambient tem-

peratures.

2. Aerosol foam compositions according 100 to claim 1 in which said surfactant includes ethers of sorbitan fatty acid esters, and ethoxylated, saturated C₁₆₋₂₀ fatty alcohols and ethoxylated lanolin alcohols.

3. Aerosol foam compositions according 105 to claim 1 in which the water-alcohol ratio is such that at ambient temperatures said surfactant blend is barely soluble therein.

4. Aerosol foam compositions according to claim 1 in which the amount of water is 110 at least 20%.

5. Aerosol foam compositions according to claim 1 in which there is present also as coupler a substance selected from the class consisting of glycols, fatty ethers and fatty 115

 Aerosol foam compositions according to claim 1 in which said aerosol foam breaks in about one minute.

7. Aerosol foam compositions according 120 to claim 1 in which said aerosol foam puffs over a range of about 6°-30° C.

8 Aerosol foam compositions according to claim I in which the propellant is a mixture of monochloro-difluoro-ethane with 125 trichloro-monofluoro-methane in about equal proportions.

9. Aerosol foam compositions according to claim 1 in which the ratios of water/alcohol/propellant/surfactant are such that 130

said foam collapses on the skin without application of pressure by the temperature of the skin. 10. Aerosol foam compositions according to claim 1 in which the following approximate composition is used:
_
Alcohol % by Wt.
Penta-ethoxy to decaethoxy-
16 Cetyl/Stearyl/Arachidyl ethers 0.40 Water

Olycellie 10.00
diffuoro-emane.
11. Aerosol foam compositions according15 to claim 1 in which the following approximate composition is used:
Alcohol % by Wt.
- XICOHOI
Cetyl/alcohol/diethoxy stearyl
20 Guler/decaethoxy stearul ether 27
Water
UICHIOTO-MONOfilioto-methana , and
cinoro-diluoro-ethane
12. Aerosol form companie:
25 to claim 1 in which the following approxi-
the following approxi-

mate composition is used:		
Alcohol Cetyl alcohol/pentethoxy fatty alcohol ether	% by Wt. 38.70	
Water Polyethylene glycol 200 monochloro-difluoro-ethane. 13. Aerosol from comments	1.90 45.90 9.70 , and	30
mate composition is used: Alcohol Diethoxy stearyl ether/correl	ing approxi- by Wt. 47.8	35
alcohol/polyethylene (25) glycol lanolin alcohol ether Water trichloro-monofluoro-methane v chloro-difluoro-ethane.	2.9 38.6 , and with mono-	40

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Printed for Her Majesty's Stationery Office by The Tweeddale Press Ltd., Berwick-upon-Tweed, 1968.

Published at the Patent Office, 25 Southampton Buildings, London, W.C.2, from which copies may be obtained.

